

AF 9362

(19)



Europäisches Patentamt
European Patent Office
Office européen des brevets



(11)

EP 0 683 181 B1

(12)

EUROPEAN PATENT SPECIFICATION

(45) Date of publication and mention
of the grant of the patent:
10.09.1997 Bulletin 1997/37

(51) Int Cl.⁶: **C08F 16/32, C07C 43/313**

(21) Application number: **95107193.5**

(22) Date of filing: **12.05.1995**

(54) **Fluorinated polymers and copolymers containing cyclic structures**

Fluorierte Polymere und Copolymere die cyclische Strukturen enthalten

Polymères fluorés et copolymères contenant des structures cycliques

(84) Designated Contracting States:

AT BE CH DE DK ES FR GB GR IE IT LI NL PT SE

(30) Priority: **19.05.1994 IT MI941011**

(43) Date of publication of application:
22.11.1995 Bulletin 1995/47

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(56) References cited:
WO-A-90/15043
US-A- 4 910 276

US-A- 3 418 302

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Description

The present invention relates to fluorinated polymers and copolymers containing cyclic structures in the main chain. Specifically the present invention relates to fluorinated polymers and copolymers containing cyclic structures obtained by cyclopolymerization of dienes.

The polymers of the invention do not show gelling phenomena during the process for preparing them. This is important, as a matter of fact the possible formation of gels would lead to polymers having mechanical, processability and solubility characteristics unsuitable to some applications. On the other hand, the separation of gels from polymeric crude products is difficult and not always possible.

The invention relates therefore in particular to fluorinated polymers and copolymers, containing said cyclic structures, thermoprocessable and having improved thermostability combined with high solubility and absence of gels.

Said polymers and copolymers can be used in particular in preparing coatings for applications at high temperatures, where it is necessary to have high thermal stability and also high solubility to obtain solutions having high concentration of polymer.

It is known in the art that polymers containing cyclic structures in the main chain can be obtained by radical polymerization of non conjugated dienes. In particular the cyclopolymerization of fluorinated non conjugated dienes is described in US Patents 3418302, 4910276 and 5260492.

In US 3418302 it is stressed in general that the formation of not crosslinked saturated linear polymers from compounds containing two double bonds is extremely difficult and in particular how it is possible in the case of perfluorodimethylenebisvinylether with formation of cyclic structures in the main chain only on condition that the monomer is very diluted in the polymerization step, its concentration having to be lower than 12% by weight of the total monomers and diluents. In fact if the concentration of the dienic monomer is higher, gel phenomena occur which lead to the above mentioned drawbacks.

To avoid the gel formation it was suggested in the art to resort to the use of monomers having particular molecular structures.

For instance, in US 4910276 it is described the cyclopolymerization of divinyl fluorinated monomers also at high concentration of the same during polymerization, but a necessary condition for its occurrence is that the monomers have two vinylic groups with different reactivity, in practice said groups must have different structure. For the homopolymers according to this invention the thermostability data show a maximum temperature of 475°C in relation to a weight loss of 10% because of thermal effect, while as to solubility the highest value is 7%.

Likewise US Patent 5260492 describes the cyclopolymerization of non conjugated fluorinated dienes, in particular of the ω -alkenyl-vinylethers type, but also in this case monomers having unsaturated groups with different reactivity are used.

Also polymers and copolymers obtained according to the last patents do not show in general an high thermal stability. In the case of homopolymers there is indeed a thermostability just slightly higher than that of the polymers according to US patent 4910276: the maximum attainable temperature before losing 10% by weight because of thermal effect is about 485°C.

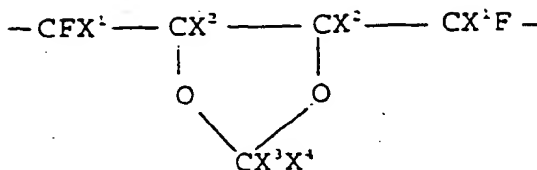
Moreover, in the preparation of polymers according to said patents one has, to a different extent, the rearrangement of the vinylether group with formation of acylfluoride -COF, which, as known in the art, leads to a reduction of the thermal stability of the polymers themselves. If it is desired to reduce the content of -COF of said polymers, one could resort to a subsequent fluorination process, according to what known in the art, such resort, however, would make more complex the preparation of the polymers themselves.

It has now been surprisingly and unexpectedly found that it is possible to obtain by diene cyclopolymerization linear fluorinated polymers thermally more stable than those of the prior art, the content in fluorine being same, free from gels and endowed with high solubility, if a particular class of non conjugated fluorinated dienes is used. Such dienes, contrary to what reported in US patent 4910276, cyclopolymerize even though they have two vinylic groups with the same reactivity. Moreover, unexpectedly, with the monomers of the invention, rearrangement phenomena and formation of -COF groups do not occur.

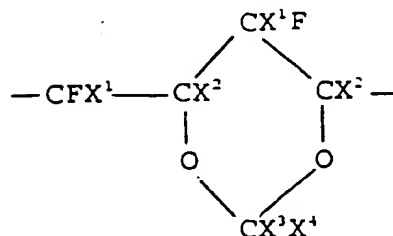
According to the present invention it has surprisingly and unexpectedly found that fluorinated polymers and copolymers with cyclic structures in main chain having said characteristics, can be obtained by using, as diene, bisvinylloxymethanes having the structure $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2=\text{CX}^1\text{F}$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 .

Object of the present invention are therefore polymers and copolymers derived from one or more of said bisvinylloxymethanes forming essentially only repetitive cyclic units having the structures a) and b):

a)

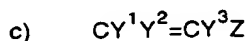


b)



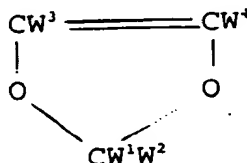
wherein X^1 , X^2 , X^3 and X^4 have the meaning indicated above.

In the case of copolymers, at least one of the other comonomers is a compound with an ethylene unsaturation. Utilizable comonomers of this type are olefins, optionally containing also chlorine and/or fluorine atoms, alkylvinylethers and dioxolic compounds. More precisely olefins have the structure of the type



wherein Y^1 , Y^2 , Y^3 , equal to or different from each other, are F, Cl or H; Z is F, Cl, H or a perfluoroalkyl radical R^f containing from one to five carbon atoms, with the proviso that the olefin contains 2 chlorine atoms at most and preferably no more than one chlorine atom; alkylvinylethers have the same structure c) wherein Z is O-R^f with R^f equal to a perfluoroalkyl radical having from one to five carbon atoms, with the same limitations relating to the content of chlorine atoms. The dioxolic compounds have the structure:

d)



wherein W^1 and W^2 , equal to or different from each other, represent F or CF_3 , W^3 represents F or H, and W^4 represents F, H, R^f or O-R^f with R^f equal to a perfluoroalkyl radical having from 1 to 5 carbon atoms. The preparation of said dioxolic compounds is described in US patents 4908461, 5245054, 5296617 when W^4 is different from R^f and O-R^f . When W^4 is R^f or O-R^f , for different dioxoles than the ones cited in said US patents see the European patent application EP-A-633257.

Preferred polymers and copolymers are those derived from monomers with $X^2=X^1$, having the structure $\text{CFX}^1=\text{CX}^1\text{---O---CX}^3\text{X}^4\text{---O---CX}^1=\text{CX}^1\text{F}$, wherein X^1 is F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 ; more preferred polymers and copolymers are those derived from monomers having the structure $\text{CF}_2=\text{CF---O---CX}^3\text{X}^4\text{---O---CF}=\text{CF}_2$, wherein X^3 and X^4 , equal or different from each other, are F or CF_3 and even more preferred are those derived from perfluorobisvinylloxymethane $\text{CF}_2=\text{CF---O---CF}_2\text{---O---CF}=\text{CF}_2$.

As comonomers, one or more of the following compounds can be preferably used: tetrafluoroethylene, vinylidene fluoride, ethylene, chlorotrifluoroethylene, trifluoroethylene, perfluoromethylvinylether, perfluoropropylvinylether, perfluoropropene, perfluoro 2,2-dimethyl-1,3-dioxole, perfluoro 1,3-dioxole, 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole.

The obtained polymers show the combination of the aforesaid cyclic structures in main chain, the 5 atoms ring structures being prevalent, while both open structures containing double bonds in side chain and carbonylic groups deriving from the rearrangement of the perfluorovinylether group are substantially absent.

The polymers and copolymers according to the present invention, besides being chemically stable, have a surprising thermal stability, high transparence and in some cases, even though they have an high content in fluorine, are soluble in some common solvents, such as for instance acetone, diethyleneglycoledimethylether and N,N-dimethylformamide. The perfluorobisvinylloxymethane homopolymer and some copolymers according to this invention show on the other hand a good solubility in the fluorinated solvents, such as hydrochlorofluorocarbons, fluorohydrocarbons, perfluoroamines, perfluoroethers, similar solvents and their mixtures and high solubility in Fluorinert^(R) FC 75 (perfluoro-2-butyl-tetrahydrofuran) and in perfluoropolyethers Fomblin^(R) and Galden^(R) having, optionally, one or more H in one or in both the ending groups (e.g. -CF₂H).

In particular, the perfluorobisvinylloxymethane homopolymer according to this invention, has a weight loss of 10% only at a temperature of about 530°C and it shows an high solubility in the perfluoropolyethers Fomblin^(R) and Galden^(R) and in Fluorinert^(R) FC 75.

These properties make such polymers and copolymers very suitable to be used also for certain applications as protective coating both of electric wires and particularly of metals, in applications of optical type, particularly in the manufacture of optical fibers, where the low refraction index and the high transparence of these products are used also in the area of wave lengths corresponding to the ultraviolet, and in the preparation of manufactured articles where high thermal stability is required.

The polymers and copolymers of the present invention can be prepared by radical polymerization in solution or in absence of solvent, in suspension or in mass.

General polymerization methodologies utilizable in a non aqueous medium are described for instance in US Patents 4864006 and 5182342, while methodologies utilizable in an aqueous medium are described in EP Patent 247379 and again in US patent 5182342. Any substance capable of generating radicals under the chosen reaction conditions can be used as polymerization initiator, in particular perfluoropropionylperoxide, benzoylperoxide, azobisisobutyronitrile or percarbonates can be used. It is also possible to start the polymerization using systems of redox couples, for instance those described in Prog. Polym. Sci., vol. 8, page 61, 1982.

The fluorinated derivatives of bisvinylloxymethane having the structure $CFX^1=CX^2-O-CX^3X^4-O-CX^2'=CX^1'F$, where- in X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF₃; X^1' and X^2' , equal to or different from each other, are X^1 or X^2 , with the proviso that when X^1 is different from X^2 also X^1' is to be different from X^2' , are new as such and are obtainable according to an innovative process. Among these compounds, those wherein $X^1=X^1'$ and $X^2=X^2'$ cyclopolymerize without giving -COF groups in the polymer; this occurs in particular in case of perfluorobisvinylloxymethane $CF_2=CF-O-CF_2-O-CF=CF_2$.

A further object of the present invention is therefore the process for preparing fluorinated derivatives of bisvinylloxymethane having the structure $CFX^1=CX^2-O-CX^3X^4-O-CX^2'=CX^1'F$, wherein X^1 and X^2 equal to or different from each other are F, Cl or H, X^3 and X^4 equal to or different from each other are F or CF₃, X^1' and X^2' equal to or different from each other are X^1 or X^2 , with the proviso that when X^1 is different from X^2 also X^1' is different from X^2' , which comprises:

- i) addition of an olefin of formula $CX^1X^5=CX^2X^6$, wherein X^1 , X^2 , X^5 and X^6 equal to or different from each other are F, Cl, H or Br, the Br atoms being 2 at most and in such a case bound to different carbon atoms, X^1 and X^5 not being both F, X^2 and X^6 not being both F, X^1 and X^2 being F only if X^5 and X^6 are different from F and not being both H, in a reactor where an hypofluorite of general formula $CX^3X^4(OF)_2$ wherein X^3 and X^4 equal to or different from each other are F or CF₃, is essentially always present, dissolved in an inert solvent having a concentration comprised between 0.001 M and 10 M, at a temperature from -140°C to +60°C;
- ii) separation of the reaction product between two olefin molecules and one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.
- iii) dehalogenation or dehydrohalogenation of the product obtained in ii), where the eliminated halogen atoms are Cl or Br.

The Applicant has surprisingly found experimental conditions such as to substantially and not predictably modify the reactivity of the hypofluorites having the general formula $CXY(OF)_2$ towards the olefins when one operates in solution, succeeding in obtaining with good yield linear reaction products between one molecule of hypofluorite and two molecules of olefin.

According to US Patents 5225576 and 5235074, such reactivity leads in fact essentially to the formation of a 1,3-dioxolane derivative and to the fluorination product of the olefin itself.

Linear addition products between hypofluorite and olefins in the presence of solvent are described in Inorganic Chemistry vol. 7, No. 3, 1968, page 624-6, where the hypofluorite and the olefin are first condensed in stoichiometric ratio at -184°C and by subsequent heating they are reacted in absence of an excess of hypofluorite, conditions under which the preferred product is not the linear product but the dioxolanic compound, as described in the aforesaid patents.

The experimental conditions which surprisingly allow to obtain the linear addition product as preferred product

refer to the addition methodology of the reactants used and to the polarity of the reaction medium.

It has been surprisingly found as a matter of fact that by adding the olefin to the hypofluorite dissolved in a solvent, provided that it does not react with the hypofluorite, with a concentration from 0.001 M to 10 M, preferably from 0.1 M to 4 M, at a temperature from -140°C to +60°C and preferably from -120°C to 0°C, the reaction proceeds with a clear increase of the formation of the aforesaid linear reaction product. The inert solvent can be an ordinary one with low polarity, for instance dichlorodifluoromethane, trichlorofluoromethane, Fluorinert^(R) FC 75 or a perfluoropolyether Fomblin^(R) and Galden^(R), but preferably the solvent used is a polar one. The polar solvent is preferably selected from hydrogenfluorocarbons, hydrogenchlorocarbons, fluorochlorocarbons, hydrogenchlorofluorocarbons, trifluoroacetic acid, trifluoroacetic anhydride, acetic nitrile, hydrofluoric acid, sulphur dioxide, trifluoromethanesulphonic acid, CF₂Cl-CFCl-SO₂F, mixtures of the same and mixtures of one or more of said solvents also in small percentage with an ordinary solvent having low polarity. Among chlorinated and fluorochlorinated solvents the ones containing also hydrogen in the molecule are preferred, because of their lower impact on the decrease in the atmosphere ozone layer.

The olefin used in the process is selected depending on the desired intermediate or dienic compound. Preferred olefins are CFCI=CFCl, CHCl=CHCl, CHCl-CCl₂, CCl₂=CCl₂, CH₂=CF₂, CF₂=CF₂, CFH=CFCl, CFCI=CHCl, CH₂=CCl₂, CH₂=CFCl.

The process can be carried out both in a continuous way, introducing the olefin in a reactor where hypofluorite is always present at a concentration kept constant and in a discontinuous way by adding the olefin to an hypofluorite solution in a suitable solvent without restoring the concentration of the hypofluorite which progressively reacts.

The dehalogenation or dehydrohalogenation of the product obtained in the phase ii) of the process can be carried out according to one of the methods described in the art.

The process is preferably used for the preparation of bisvinyloxymethane derivatives having the structure CFX¹=CX¹-O-CX³X⁴-O-CX¹=CX¹F, wherein X¹ is F, Cl or H; X³ and X⁴, equal to or different from each other, are F or CF₃, using an olefin of formula CX¹X⁵=CX¹X⁶, where X¹, X⁵ and X⁶, equal to or different from each other, are F, Cl, Br or H, X¹ and X⁵ not being both F, X¹ and X⁶ not being both F, X¹ being Br only when X⁵ and X⁶ equal to each other are different from Br, X¹ being F only when X⁵ and X⁶ are different from F and not being both H.

The process is used even more preferably for the preparation of bisvinyloxymethane derivatives having the structure CF₂=CF-O-CX³X⁴-O-CF=CF₂, wherein X³ and X⁴, equal to or different from each other, are F or CF₃, using an olefin of formula CFX⁵=CFX⁶, wherein X⁵ is Cl, Br or H, and X⁶ is Cl or Br.

It has been found, moreover, that if the solvent used in the reaction between olefin and hypofluorite is of polar type, the addition sequence of the reactants can be inverted. A further object of the present invention is therefore a process for preparing fluorinated derivatives of bisvinyloxymethane having the structure CFX¹=CX²-O-CX³X⁴-O-CX²=CX¹F, wherein X¹ and X², equal to or different from each other, are F, Cl or H; X³ and X⁴, equal to or different from each other, are F or CF₃; X¹ and X², equal to or different from each other, are X¹ or X², with the proviso that if X¹ is different from X² also X¹ is to be different from X², comprising:

i) addition of hypofluorite having the general formula CX³X⁴(OF)₂ wherein X³ and X⁴, equal to or different from each other, are F or CF₃ to an olefin of formula CX¹X⁵=CX²X⁶, wherein X¹, X², X⁵ and X⁶, equal to or different from each other, are F, Cl, H or Br; the Br atoms being 2 at most and in such a case bound to different carbon atoms; X¹ and X⁵ not being both F; X² and X⁶ not being both F; X¹ and X² being F only if X⁵ and X⁶ are different from F and not being both H; said olefin being dissolved in an inert solvent of polar type, at a temperature from -140°C to +60°C, preferably from -120°C to 0°C;

ii) separation of the reaction product of two olefin molecules with one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), where the eliminated halogen atoms are Cl or Br.

Preferably the inert solvent of polar type is selected from hydrogenfluorocarbons, hydrogenchlorocarbons, fluorochlorocarbons, hydrogenchlorofluorocarbons, trifluoroacetic acid, trifluoroacetic anhydride, acetic nitrile, hydrofluoric acid, sulphur dioxide, trifluoromethanesulphonic acid, CF₂Cl-CFCl-SO₂F, mixtures of the same and mixtures of one or more of said solvents also in small percentage with an ordinary low polarity solvent.

Also for this process the preferred olefins are CFCI=CFCl, CHCl=CHCl, CHCl-CCl₂, CCl₂=CCl₂, CH₂=CF₂, CF₂=CF₂, CFH=CFCl, CFCI=CHCl, CH₂=CCl₂, CH₂=CFCl.

The dehalogenation or dehydrohalogenation of the product obtained in step ii) of the process can be likewise carried out according to one of the methods described in the art.

Like the previous process, also this process is preferably used for the preparation of bisvinyloxymethane derivatives having the structure CFX¹=CX¹-O-CX³X⁴-O-CX¹=CX¹F, wherein X¹ is F, Cl or H; X³ and X⁴, equal to or different from each other, are F or CF₃, by using an olefin of formula CX¹X⁵=CX¹X⁶ wherein X¹, X⁵ and X⁶ equal to or different from each other are F, Cl, Br or H, X¹ and X⁵ not being both F, X¹ and X⁶ not being both F, X¹ being Br only when X⁵ and

X^6 equal each other are different from Br, X^1 being F only when X^5 and X^6 are different from F and not being both H.

Likewise also this process is even more preferably used for preparing bisvinylmethane derivatives having the structure $CF_2=CF-O-CX^3X^4-O-CF=CF_2$, wherein X^3 and X^4 , equal to or different from each other, are F or CF_3 , using an olefin of formula $CFX^5=CFX^6$, wherein X^5 is Cl, Br or H, and X^6 is Cl or Br.

The fluorinated derivatives of bisethoxymethane obtainable after steps i) and ii) of the processes described above, possibly as a mixture of more compounds, are a further object of the present invention. These products, besides being used as intermediates for the synthesis of monomers, are also used as fluorinated solvents with low impact on the atmospheric ozone layer. The products without chlorine are preferred for this use.

The following examples are to be considered for illustrative purposes, but not limitative of the scope of the present invention.

EXAMPLES 1-16 Bisethoxymethane derivatives from olefin and hypofluorite: addition of the hypofluorite to the olefin in solvents having different polarity.

EXAMPLE 1

80 g (600 mmoles) of 1,2-dichloro-1,2-difluoroethylene and a solvent mixture formed by 32 g of SO_2 and 17 g of CH_2Cl_2 are introduced into a 125 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture and inert gas head outlet.

The so loaded reactor is brought to the temperature of $-80^\circ C$ using a cryostat, and then a mixture of $CF_2(OF)_2$ (1,5 NI/h), CO_2 (2,2 NI/h) and helium (6 NI/h) are continuously added under mechanical stirring for 3.5 hours with an overall supply of hypofluorite equal to 234 mmoles.

The reaction is practically instantaneous. When the addition of the mixture containing hypofluorite is over, the products are separated from the reaction crude product by fractional distillation with a plate column at atmospheric pressure. The fractions having boiling point in the range comprised between -1° and $+1^\circ$ with respect to the temperatures indicated hereinafter, are collected:

a) at $+4^\circ C$ 19.3 g (112 mmoles) of CF_2Cl-CF_2Cl

b) at $+22^\circ C$ 17.4 g (131 mmoles) of unreacted $CFCI=CFCl$

c) at $47/49^\circ C$ mixture formed by 24.4 g (113 mmoles) of 4,5-dichloro-2,2,4,5-tetrafluoro-1,3-dioxolane and 4 g (22 mmoles) of $CClF_2-CCl_2F$

d) at $+76^\circ C$ 8.6 g (40 mmoles) $CF_2Cl-CFCI-OC(O)F$

e) at $+70.5^\circ C/54$ mm Hg 20.4 g (52 mmoles) of $CF_2Cl-CFCI-OCF_2O-CFCI-CF_2Cl$.

The linear addition product yield, defined as ratio between the obtained moles of $CF_2Cl-CFCI-OCF_2O-CFCI-CF_2Cl$ and the used moles of $CF_2(OF)_2$, is 22%. The product has been characterized by F^{19} -NMR spectroscopy (in p.p.m. related to $CFCI_3 = O$: 2F type $O-CF_2-O$ at $-50.5/-52.0$; 4F type $Cl-CF_2-C$ at $-70.3/-72.3$; 2F type $C-CFCI-O$ at -77.5), and electronic impact mass spectroscopy, where the main peaks and the relative intensities are: 151 (100%), 101 (23%), 85 (15%), 66 (12%), 47 (12%), 28 (33.5%).

EXAMPLES 2-11

In these examples some solvents and co-solvents are tested at different temperatures under experimental conditions comparable with the ones of example 1. The data relating to the yield in linear addition product formed by two molecules of olefin and one molecule of hypofluorite and under the experimental conditions adopted are reported in Table 1.

EXAMPLE 12

In this example the use of a different olefin, the trans-1,2-dichloroethylene, is tested.

174 g (1,79 moles) of trans-1,2-dichloroethylene and a solvent mixture formed by 118 g of SO_2 and 247 g of $CFCI_3$ are introduced into a 500 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture and inert gas head outlet.

The so loaded reactor is brought to the temperature of $-76^\circ C$ using a cryostat, and then a mixture of $CF_2(OF)_2$ (2 NI/h), CO_2 (1.0 NI/h) and helium (6 NI/h) are continuously added under mechanical stirring for 8.5 hours with an overall supply of hypofluorite equal to 760 mmoles.

When the addition of hypofluorite is over, the sulphur dioxide and part of $CFCI_3$ are separated from the reaction crude product by distillation. Then the reaction products are separated by fractional distillation with a plate column at

reduced pressure. The fractions having boiling point in the range between -1° and $+1^{\circ}$ with respect to the temperature indicated hereinafter, are collected:

- a) at $+20^{\circ}\text{C}/70$ mm Hg mixture of 50 g of CFCIH-CFCIH and 25 g of unreacted trans-1,2-dichloroethylene;
- b) at $+29^{\circ}\text{C}/58$ mm Hg 46 g of 4,5-dichloro-2,2-difluoro-1,3-dioxolane;
- c) at $+12^{\circ}\text{C}/24$ mm Hg 20 g of CFCIH-CCIH-OC(O)F
- d) at $+64^{\circ}\text{C}/3$ mm Hg 71 g of $\text{CFCIH-CHCI-OCF}_2\text{O-CHCI-CFCIH}$.

The linear addition product yield, defined as ratio between the obtained moles of $\text{CFCIH-CHCI-OCF}_2\text{O-CHCI-CFCIH}$ and the used moles of $\text{CF}_2(\text{OF})_2$, is 30%. The product has been characterized by F^{19} -NMR spectroscopy (in p.p.m. related to $\text{CFCI}_3 = \text{O}$: 2F type $\text{O-CF}_2\text{-O}$ at $-58.0/-61.5$; 2F type CHFCl-CHCl at $-141/-146$), H^1 -NMR spectroscopy (in p.p.m. related to TMS = 0: complex multiplet at $+5.6/6.9$) and mass spectroscopy (main peaks: 209, 211, 213 for $\text{C}_4\text{H}_3\text{OCl}$; 179, 181, 183 for $\text{C}_3\text{H}_2\text{OF}_3\text{Cl}_2$; 131, 133, 135 for $\text{C}_2\text{H}_2\text{OFCI}$; 15, 117, 119 for CH_2FCI_2).

EXAMPLES 13 and 14

In these examples the use of trans-1,2-dichloroethylene is tested, with the same procedure of example 12, under different conditions reported in Table 1 along with the linear addition product yield.

EXAMPLE 15

79 g (0.6 moles) of trichloroethylene and a solvent mixture formed by 39 g of SO_2 and 83 g of CFCl_3 are introduced into a 500 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture and inert gas head outlet.

The so loaded reactor is brought to the temperature of -74°C using a cryostat, and then a mixture of $\text{CF}_2(\text{OF})_2$ (1 N/h), CO_2 (0.5 N/h) and helium (3 N/h) are continuously added under mechanical stirring for 6 hours.

When the addition of hypofluorite is over, solvents and most volatile reaction products CHFCl-CFCl_2 and 4,4,5-trichloro-2,2,4-trifluoro-1,3-dioxolane are stripped off the crude reaction mixture.

33 g of isomeric, essentially pure bisethereal addition products, having molecular weight 382.8, are left in the kler. The bisethereal products can be separated by gaschromatography. The characterization is reported in example 18.

The linear addition product yield, defined as ratio between the moles of obtained bisethereal products and the used moles of $\text{CF}_2(\text{OF})_2$, is 32%.

EXAMPLE 16

100 g (0.6 moles) of tetrachloroethylene and a solvent mixture formed by 39 g of SO_2 and 124 g of CFCl_3 are introduced into a 500 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture and inert gas head outlet.

The so loaded reactor is brought to the temperature of -54°C using a cryostat, and then a mixture of $\text{CF}_2(\text{OF})_2$ (1 N/h), CO_2 (0.5 N/h) and helium (3 N/h) are continuously added under mechanical stirring for 6 hours.

When the addition of hypofluorite is over, solvents and most volatile reaction products $\text{CFCl}_2\text{-CFCl}_2$ and 4,4,5-tetrachloro-2,2-difluoro-1,3-dioxolane are stripped off the crude reaction mixture.

32 g of the essentially pure bisethereal addition product $\text{CFCl}_2\text{-CCl}_2\text{-OCF}_2\text{O-CCl}_2\text{-CFCl}_2$ are left in the kler.

The linear addition product yield, defined as ratio between the moles of obtained bisethereal product and the used moles of $\text{CF}_2(\text{OF})_2$, is 26%.

The product $\text{CFCl}_2\text{-CCl}_2\text{-OCF}_2\text{O-CCl}_2\text{-CFCl}_2$ is characterized by F^{19} -NMR spectroscopy, where it shows in p.p.m., related to $\text{CFCl}_3 = \text{O}$, 2F type OCF_2O at -50.7 and 2F type CFCl_2 at -67.5 .

The mass spectroscopy analysis shows the following main peaks and relative intensities: 185 (100%), 148 (15%), 101 (12%).

TABLE 1

Formation of linear addition product between hypofluorite and olefin: addition of hypofluorite to the olefin ^(*)						
Ex.	Solvent (g)		Cosolvent (g)		Temperature (°C)	Linear addition product yield (% by moles)
1	SO ₂	32	CH ₂ Cl ₂	17	-80	22.0
2	CH ₂ Cl ₂	8			-90	10.5
3	SO ₂	32			-80	18.6
4	SO ₂	19.2			-80	17.5
5	SO ₂	19.2			-90	8.5
6	SO ₂	19.2			-60	17.0
7	SO ₂	64	CH ₂ Cl ₂	34	-80	18.3
8	CF ₂ Cl-CFCl-SO ₂ F	127			-80	18.5
9	HF	8			-80	12.4
10	(CF ₃ C(O)) ₂ O	8			-70	12.0
11	CF ₃ SO ₃ H	8			-55	11.0
12	SO ₂	118	CFCl ₃	247	-76	30.0
13	SO ₂	118	CH ₂ Cl ₂	153	-70	43.0
14	-		CHCl ₂ -CF ₃	250	-70	36.0
15	SO ₂	39	CFCl ₃	83	-74	32.0
16	SO ₂	39	CFCl ₃	124	-54	26.0

(*) The hypofluorite is CF₂(OF)₂; the olefin is 1,2-dichloro-difluoroethylene in the examples from 1 to 11, trans 1,2-dichloroethylene in the examples from 12 to 14, trichloroethylene in example 15 and tetrachloroethylene in example 16.

EXAMPLES 17-22: Bisethoxymethane derivatives from olefin and hypofluorite: addition of olefin to hypofluorite.

EXAMPLE 17

2 ml of CFCl₃ are loaded in a 10 ml two necks glass flask equipped with magnetic stirrer and internal thermocouple. The reactor is brought to -105°C and evacuated. Subsequently by vacuum line transfer 1 mmole of CF₂(OF)₂ is condensed therein at -196°C. The so loaded reactor is brought to -100°C and the solution of CF₂(OF)₂ in CFCl₃ is homogenized under magnetic stirring. 3 mmoles of CFCl=CFCl are slowly added by condensation to the so obtained 0,5 molar solution of CF₂(OF)₂ in CFCl₃ and maintained at -100°C so that the reaction heat immediately developed is effectively vented by the cooling system.

The weight percentages of the reaction products, determined by gaschromatography and by NMR, are: CF₂Cl-CF₂Cl 40.3%; 4,5-dichloro-2,2,4,5-tetrafluoro-1,3-dioxolane 50.7%; CF₂Cl-CFCl-OCF₂O-CFCl-CF₂Cl 9%. The linear addition product, defined as for example 1, is 9%.

EXAMPLES 18 and 19

Under experimental conditions comparable with those of example 17, some solvents and CF₂(OF)₂ concentrations are tested at different temperatures. The data relating to the linear adduct yield and to the adopted experimental conditions are reported in Table 2.

EXAMPLE 20

20 ml of CHCl₂-CF₃ (d=1.5 g/l) are loaded into a 125 ml multi-neck glass cylindrical reactor, equipped with me-

chanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture, inlet on the reactor top for the liquid reacting olefin, inert gas head outlet. The so loaded reactor is brought to -97°C by means of a cryostat, and then a gaseous mixture of $\text{CF}_2(\text{OF})_2$ (0.50 Nl/h), CO_2 (0.25 Nl/h) and He (1.5 Nl/h) is continuously introduced under mechanical stirring until a 1.5 M concentration of $\text{CF}_2(\text{OF})_2$ in solution is achieved.

At this point $\text{CHCl}=\text{CCl}_2$ in admixture with hypofluorite is continuously added so as to keep constant the concentration of the latter in the solution during the addition. After 4h and 30', the addition of the gaseous mixture containing $\text{CF}_2(\text{OF})_2$ is interrupted and the addition of $\text{CHCl}=\text{CCl}_2$ is continued up to the complete conversion of $\text{CF}_2(\text{OF})_2$ dissolved in the solution.

At the end of the reaction, the reaction crude product is stripped from the solvent and from the most volatile reaction by-products: $\text{CHFCl}-\text{CFCl}_2$ and 4,4,5-trichloro-2,2,4-trifluoro-1,3-dioxolane. 34.5 g of a mixture of bisethereal addition products formed by:

- A) 71.5% of $\text{CCl}_2\text{F}-\text{CHCl}-\text{OCF}_2\text{O}-\text{CHCl}-\text{CCl}_2\text{F}$
- B) 27.7% of $\text{CCl}_2\text{F}-\text{CHCl}-\text{OCF}_2\text{O}-\text{CCl}_2-\text{CHClF}$
- C) 0.8% of $\text{CHClF}-\text{CCl}_2-\text{OCF}_2\text{O}-\text{CCl}_2-\text{CHClF}$ are obtained as residue.

The products are separated by preparative gaschromatography. The linear addition product yield, defined as ratio between the sum of the moles of products A, B and C and the moles of $\text{CF}_2(\text{OF})_2$ converted, is 69%.

The products are characterized by F^{19} -NMR spectroscopy where they show, in p.p.m., related to $\text{CFCl}_3 = \text{O}$:

- product A: 2F type CFCl_2 at -66.5; 2F type OCF_2O at -59.3/-61.5;
- product B: 1F type CFCl_2 at -66.2; 1F type CFHCl at -138.3; 2F type OCF_2O at -55.0/-56.3;
- product C: 2F type CFHCl at -138.4; 2F type OCF_2O at -50.6.

The mass spectroscopy analysis shows the following main peaks and relative intensities: 151 (100%), 131 (23.5%), 114 (33%), 101 (20%), 79 (14%).

EXAMPLE 21

54 ml of $\text{CH}_2\text{F}-\text{CF}_3$ ($d=1.2$ g/l) are loaded into a 125 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture $\text{CF}_2(\text{OF})_2/\text{CO}_2$ /diluent gas, suction inlet for the reacting olefin, inert gas head outlet. The so loaded reactor is brought to -90°C by means of a cryostat, and then a gaseous mixture of $\text{CF}_2(\text{OF})_2$ (1.0 Nl/h), CO_2 (0.5 Nl/h) and He (4 Nl/h) is continuously introduced under mechanical stirring until a 0.2 M concentration of $\text{CF}_2(\text{OF})_2$ in solution is achieved.

At this point $\text{CClF}=\text{CHF}$ (2.0 Nl/h) is continuously added so as to keep constant the concentration of the hypofluorite in the solution during the addition. After 1h and 40', the addition of the gaseous mixture containing $\text{CF}_2(\text{OF})_2$ is interrupted and the addition of $\text{CClF}=\text{CHF}$ is continued up to the complete conversion of $\text{CF}_2(\text{OF})_2$ dissolved in the solution.

At the end of the reaction, the reaction crude product is stripped from the solvent and from the most volatile reaction by-products: $\text{CHF}_2-\text{CF}_2\text{Cl}$ and 4-chloro-2,2,4,5-tetrafluoro-1,3-dioxolane. 16.5 g of a mixture of bisethereal addition products formed by:

- A) 74% of $\text{CClF}_2-\text{CHF}-\text{OCF}_2\text{O}-\text{CHF}-\text{CClF}_2$
- B) 23% of $\text{CClF}_2-\text{CHF}-\text{OCF}_2\text{O}-\text{CClF}-\text{CHF}_2$
- C) 3% of $\text{CHF}_2-\text{CClF}-\text{OCF}_2\text{O}-\text{CClF}-\text{CHF}_2$

are obtained as residue.

The products are separated by preparative gaschromatography. The linear addition product yield, defined as ratio between the sum of the moles of products A, B and C and the moles of $\text{CF}_2(\text{OF})_2$ converted, is 61%. The products are characterized by F^{19} -NMR spectroscopy and H^1 -NMR spectroscopy

By F^{19} -NMR spectroscopy the products show, in p.p.m., related to $\text{CFCl}_3 = \text{O}$:

- product A: 4F type CF_2Cl at -71.5; 2F type CHF at -140.3;
2F type OCF_2O at -58.5/-60.0;
- product B: 2F type CF_2Cl at -71.5; 1F type CHF at -140.3;
1F type CFCl at -81.2; 2F type CHF_3 at -131/-134;
2F type OCF_2O at -54.0/-56.6;
- product C: 2F type CFCl at -81.2; 4F type CHF_2 at -131/-134;
2F type OCF_2O at -54.0/56.6.

By ^1H -NMR spectroscopy the products show, in p.p.m., related to TMS = 0:

product A: 2H type CHF at 5.8-6.0
 product B: 1H type CHF at 5.8-6.0;
 1H type CF_2H at 5.7-5.9-6.1
 product C: 2H type CF_2H at 5.7-5.9-6.1

The electronic impact mass spectroscopy analysis shows the following main peaks and relative intensities: 185 (7%), 183 (20%), 119 (38%), 117 (100%), 67 (15%).

EXAMPLE 22

160 ml of $\text{CH}_2\text{F}-\text{CF}_3$ ($d=1.2$ g/l) are loaded into a 250 ml multi-neck glass cylindrical reactor, equipped with mechanical stirrer, thermocouple, suction inlet for the reacting gaseous mixture $\text{CF}_2(\text{OF})_2/\text{CO}_2$ /diluent gas, suction inlet for the reacting olefin, inert gas head outlet. The so loaded reactor is brought to -95°C by means of a cryostat, and then a gaseous mixture of $\text{CF}_2(\text{OF})_2$ (1.0 NI/h), CO_2 (0.5 NI/h) and He (4 NI/h) is continuously introduced under mechanical stirring until a 0.05 M concentration of $\text{CF}_2(\text{OF})_2$ in solution is achieved.

At this point vinylidene fluoride $\text{CH}_2=\text{CF}_2$ (2.0 NI/h) is continuously added so as to keep constant the concentration of the hypofluorite in the solution during the addition. After 2h and 40', the addition of the gaseous mixture containing $\text{CF}_2(\text{OF})_2$ is interrupted and the addition of $\text{CH}_2=\text{CF}_2$ is continued up to the complete conversion of $\text{CF}_2(\text{OF})_2$ dissolved in the solution.

At the end of the reaction, 21 g of $\text{CF}_3\text{CH}_2\text{OCF}_2\text{OCH}_2\text{CF}_3$ (b.p. $+51^\circ\text{C}/200$ mm Hg) are isolated by fractional distillation.

The linear addition product yield, defined as ratio between the moles of $\text{CF}_3\text{CH}_2\text{OCF}_2\text{OCH}_2\text{CF}_3$ and the moles of $\text{CF}_2(\text{OF})_2$ converted, is 67%. The product has $T_g = -127.9^\circ\text{C}$. The product is further characterized by F^{19} -NMR spectroscopy and mass spectroscopy.

By F^{19} -NMR spectroscopy the product shows 2F type OCF_2O at -64.0 p.p.m. and 6F type CF_3-CH_2 at -73.9 p.p.m., related to $\text{CFCl}_3 = 0$.

The electronic impact mass spectroscopy analysis shows the following main peaks and relative intensities: 179 (33.1%), 163 (28.2%), 149 (100%), 113 (17.7%), 83 (83.1%).

The linear adduct yield and the experimental conditions adopted in examples 17-22 are summarized in Table 2.

TABLE 2

Ex.	Solvent (g)	$\text{CF}_2(\text{OF})_2$ (moles/l)	Temperature ($^\circ\text{C}$)	Linear addition product yield (% by moles)
17	CFCl_3	0.5	-100	9
18	CF_3-CFH_2	2	-95	19.2
19	$\text{CFCl}-\text{CFCl}-\text{SOF}$	2	-90	26.1
20	$\text{CHCl}_2-\text{CF}_3$	1.5	-97	69
21	$\text{CH}_2\text{F}-\text{CF}_3$	0.2	-90	61
22	$\text{CH}_2\text{F}-\text{CF}_3$	0.05	-95	67

(*) The hypofluorite is $\text{CF}_2(\text{OF})_2$; the olefin is 1,2-dichloro-difluoroethylene in the examples from 17 to 19, trichloroethylene in example 20, 1-chloro-1,2-difluoro ethylene in example 21 and vinylidene fluoride in example 22.

EXAMPLE 23 Preparation of perfluorobisvinylloxymethane (dehalogenation process).

500 ml of dimethylformamide, 90 g of Zn in powder previously activated by washing with 3N HCl and 100 mg of I_2 are loaded into a 1 l three necks flask, equipped with mechanical stirrer, thermometer, dripping funnel, distillation column with water cooler and collecting trap kept at -78°C and connected to the vacuum of a mechanical pump. The internal temperature is brought to 80°C , the vacuum is adjusted at 160 mm Hg and 88.4 g (229 mmoles) of $\text{CF}_2\text{Cl}-\text{CFCl}-\text{OCF}_2\text{O}-\text{CFCl}-\text{CF}_2\text{Cl}$ are added in drops.

Once the addition is over, the reaction mixture is kept under the conditions indicated above for 30 minutes, and then the vacuum is gradually increased up to 0.5 mm Hg. After about 20 minutes the collecting flask is disconnected;

it contains 51.3 g of a mixture consisting for 89% by weight of perfluorobisvinylloxymethane $\text{CF}_2=\text{CF}-\text{O}-\text{CF}_2\text{O}-\text{CF}=\text{CF}_2$ (45.6 g equal to 187 mmoles) and for 11% by weight of the compound of hemidehalogenation $\text{CF}_2=\text{CF}-\text{O}-\text{CF}_2\text{O}-\text{CFCl}-\text{CF}_2\text{Cl}$ (5.6 g equal to 18 mmoles). After purification by distillation at atmospheric pressure with a plate column, 42 g (172 mmoles) of perfluorobisvinylloxymethane having boiling point at atmospheric pressure of 50°C are isolated. The yield is 75%.

The product was characterized by F^{19} -NMR spectroscopy (in p.p.m., related to $\text{CFCl}_3 = \text{O}$: 2F type $\text{CF}_2=\text{C}-$ at -114.0/-121.1; 1F type $\text{C}=\text{CF}-\text{O}$ at -136.1; 2F type $\text{C}-\text{CF}_2-\text{O}$ at -60.4); IR spectrum (absorption bands at 1840, 1338, 1298, 1248, 1192 cm^{-1}) and mass spectrum (electronic impact): 163 ($\text{M}^+ - \text{C}_2\text{F}_3$); 147 ($\text{M}^+ - \text{C}_2\text{F}_3\text{O}$); 135 ($\text{C}_2\text{F}_5\text{O}$); 119 (C_2F_5).

EXAMPLE 24 Preparation of bis-(2-fluorovinylloxy)-methane $\text{CFH}=\text{CH}-\text{OCF}_2\text{O}-\text{CH}=\text{CHF}$ (dehalogenation process).

50 ml of dimethylformamide, 21 g of Zn in powder previously activated by washing with 3N HCl and 10 mg of I_2 are loaded into a 150 ml three necks flask, equipped with mechanical stirrer, thermometer, dripping funnel, distillation column with water cooler and collecting trap kept at -78°C and connected to the vacuum of a mechanical pump. The internal temperature is brought to 120°C , the vacuum is adjusted at 350 mm Hg and 17.7 g (56.4 mmoles) of $\text{CFHCl}-\text{CHCl}-\text{OCF}_2\text{O}-\text{CHCl}-\text{CFHCl}$ are added in drops.

Once the addition is over, the reaction mixture is kept under the conditions indicated above for 15 hours, and then the vacuum is gradually increased up to 200 mm Hg, after about 15 minutes the collecting flask containing 18.3 g of a mixture consisting of $\text{CFH}=\text{CH}-\text{OCF}_2\text{O}-\text{CH}=\text{CHF}$ and dimethylformamide is disconnected.

After purification of the mixture by washing twice with water and by distillation at reduced pressure with plates column, 4.6 g (26.7 mmoles) of 98% pure $\text{CFH}=\text{CH}-\text{OCF}_2\text{O}-\text{CH}=\text{CHF}$ having boiling point of 42.0°C at 200 mm Hg are isolated. The yield is 47%.

The product was characterized by F^{19} -NMR spectroscopy where it shows in p.p.m., related to $\text{CFCl}_3 = \text{O}$: 2F type $\text{O}-\text{CF}_2-\text{O}$ at -62.9; -63.3; -63.7; 1F type trans $\text{F}-\text{CH}=\text{CH}-\text{O}$ at -153.5; 1F type cis $\text{F}-\text{CH}=\text{CH}-\text{O}$ at -168.5; mass spectrum with electronic impact, wherein the main peaks and the relative intensities are: 45 (73%), 78 (17%), 111 (100%), 172 (1%); and IR spectrum wherein the following main peaks are observed: 3110, 1719, 1693, 1374, 1301, 1202, 1106 and 1028 cm^{-1} .

Example 25 Preparation of perfluorobisvinylloxymethane (dehydrohalogenation process)

3 g of bisetheral addition products mixture prepared in example 21 and 100 mg of tetrabutylammonium chloride, as phase transfer catalyst, are loaded into a 100 ml three necks flask, equipped with mechanical stirrer, thermometer, dripping funnel, distillation column with water cooler and collecting trap kept at -78°C and connected to the vacuum of a mechanical pump. The internal temperature is brought to 60°C , the vacuum is adjusted at 200 mm Hg and 3 ml of a 40% KOH water solution are added in drops.

Once the addition is over, the reaction mixture is kept under the conditions indicated above for 30', and then the vacuum is gradually increased up to 100 mm Hg. After about 30' the collecting flask is disconnected; it contains 2 g of a mixture essentially consisting of perfluorobisvinylloxymethane, unreacted or only partially reacted starting product and water traces. From the crude reaction mixture, after dehydration with solid KOH, 1.3 g of perfluorobisvinylloxymethane and 0.5 g of not dehydrohalogenated products are isolated by distillation. The yield of the perfluorobisvinylloxymethane is 66%.

The product was characterized by F^{19} -NMR spectroscopy, IR spectroscopy and mass spectroscopy as in example 23.

EXAMPLE 26 Polymerization of perfluorobisvinylloxymethane without use of solvent.

The polymerization initiator consisting of 32 microliters of 6.5% by weight perfluoropropionylperoxide in $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$ and 6.4 mmoles of perfluorobisvinylloxymethane are loaded into a 31.2 ml glass reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C , evacuated, brought to room temperature and cooled again twice; at the end of this operations sequence the reactor is kept at 30°C for 1 hour under magnetic stirring. The reaction crude product so obtained looks like a viscous, transparent, colourless and homogeneous solution.

After distillation of the unreacted monomer and stripping under vacuum at the temperature of 120°C for 3 hours, 340 mg of polymer are isolated.

The infrared analysis of the obtained polymer shows the absence of absorptions in the area of the carbonyl and in the area of the fluorinated double bond.

The F^{19} -NMR analysis is in accordance with the presence of the cyclic structures reported above, structure a)

being 89% and structure b) being 11%. Neither unreacted vinyls nor acylfluorides appear.

The intrinsic viscosity of the polymer measured in Fluorinert^(R) FC 75 is 66 ml/g. A Differential Scanning Calorimetry (DSC) test carried out on the polymers does not show any melting point, therefore the polymer is amorphous. The polymer T_g determined by DSC is 77.8°C. The thermogravimetric analysis (TGA) shows a weight loss of 2% at 488°C and of 10% at 530°C.

EXAMPLE 27 Polymerization of perfluorobisvinylmethane in solution.

The polymerization initiator consisting of 50 microliters of 6.5% by weight perfluoropropionylperoxide in $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$, 16.6 mmoles of $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$ and 8.4 mmoles of perfluorobisvinylmethane are loaded into a 52.3 ml glass reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C, evacuated, brought to room temperature and cooled again twice; at the end of this operations sequence the reactor is kept at 30°C for 2 hours under magnetic stirring. The reaction crude product so obtained looks like a viscous, transparent, colourless and homogeneous solution.

After distillation of the unreacted monomer and stripping under vacuum at the temperature of 120°C for 3 hours, 450 mg of polymer are isolated.

The infrared analysis of the obtained polymer shows the absence of absorptions in the area of the carbonyl and of the fluorinated double bond.

The F^{19} -NMR analysis is in accordance with the presence of the cyclic structures reported above, structure a) being 89% and structure b) being 11%. Neither unreacted vinyls nor acylfluorides appear.

The intrinsic viscosity of the polymer measured in Fluorinert^(R) FC 75 is 56 ml/g. The solubility of this homopolymeric sample in Fluorinert^(R) FC 75 and in perfluoropolyethers Fomblin^(R) and Galden^(R) is higher than 20% by weight at room temperature. The 20% by weight solutions in these solvents are very viscous, but transparent and visually homogeneous; solutions even with concentration higher than 20% by weight are extremely viscous, but substantially homogeneous. The T_g of the polymer determined by DSC and the TGA are substantially the same as the T_g and the TGA of the polymer obtained in example 23.

EXAMPLE 28 Copolymer of perfluorobisvinylmethane with vinylidene fluoride.

3 ml of $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$, the polymerization initiator consisting of 32 microliters of 6.5% by weight perfluoropropionylperoxide dissolved in $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$, 2.20 mmoles of fluorobisvinylmethane and 7.8 mmoles of vinylidene fluoride are loaded into a 52 ml glass reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C, evacuated, brought to room temperature and cooled again twice; at the end of this operation sequence the reactor is kept at 30°C for 16 hours under magnetic stirring. The reactor is brought to the temperature of the liquid nitrogen and connected to a vacuum system kept at the pressure of 10^{-3} mbar, it is then let to reach room temperature fractionating the vapours by traps cooled at -80°C, -120°C, and -196°C.

The trap at -80°C contains only the $\text{CFCl}_2-\text{CF}_2\text{Cl}$ used as solvent. The trap at -120°C contains 6.90 mmoles of $\text{CFCl}_2-\text{CF}_2\text{Cl}$ and 0.26 mmoles of unreacted perfluorobisvinylmethane. The trap at -196°C contains 4.5 mmoles of unreacted vinylidene fluoride.

After distillation of the solvent and of the unreacted monomers and stripping of the polymer under vacuum at the temperature of 120°C for 3 hours, 670 mg of polymer are isolated.

The weight balance determined by gaschromatography analysis of the content of the traps containing the unreacted monomers allows to calculate the amount of perfluorobisvinylmethane in the polymer, which results to be 37% by moles. The polymer results soluble in N,N-dimethylformamide.

The F^{19} -NMR analysis is in accordance with the presence of the cyclic structures reported above, structure a) being 91.4% and structure b) being 8.6%. Neither unreacted vinyls nor acylfluorides appear.

The polymer T_g determined by DSC is 17.2°C. The DSC graph does not show any melting endotherm, therefore the polymer is amorphous. The TGA shows a weight loss of 2% at 412°C and of 10% at 450°C.

EXAMPLE 29 Copolymer of tetrafluoroethylene with fluorobisvinylmethane.

8 ml of $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$, the polymerization initiator consisting of 1.5 ml of 0.35% by weight perfluoropropionylperoxide in $\text{CCl}_2\text{F}-\text{CF}_2\text{Cl}$, 0.33 mmoles of perfluorobisvinylmethane and 10 mmoles of tetrafluoroethylene are loaded into a 42 ml glass reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C, evacuated, brought to room temperature and cooled again twice; at the end of this operation sequence the reactor is kept at 40°C for 8 hours under magnetic stirring. The reaction crude

product looks like a gelatinous mass. The reactor is brought to the temperature of the liquid nitrogen and connected to a vacuum system kept at the pressure of 10^{-3} mbar, it is then let to reach room temperature fractionating the vapours by traps cooled at -80°C , -120°C , and -196°C .

The trap at -80°C contains only the $\text{CFCl}_2\text{-CF}_2\text{Cl}$ used as solvent. The trap at -120°C contains 4.90 mmoles of $\text{CFCl}_2\text{-CF}_2\text{Cl}$ and 0.16 mmoles of unreacted perfluorobisvinylloxymethane. The trap at -196°C contains 0.12 mmoles of unreacted tetrafluoroethylene.

After distillation of the solvent and of the unreacted monomers and stripping of the polymer under vacuum at the temperature of 120°C for 3 hours, 1.030 g of polymer are isolated.

The weight balance made by gas chromatography analysis of the content of the traps containing the unreacted monomers allows to calculate the amount of perfluorobisvinylloxymethane in the polymer, which results to be 1.7% by moles.

The infrared analysis of the obtained polymer shows the absence of absorptions in the area of the carbonyl and of the fluorinated double bond.

The ΔH and the second melting point determined by DSC are respectively 9.8 cal/g and 314.5°C .

The TGA shows a weight loss of 2% at 502°C and of 10% at 540°C . The polymer is hot moulded, at the temperature of 330°C and at the pressure of 122 atm in a transparent and tough film.

EXAMPLE 30 Copolymer of perfluorobisvinylloxymethane with perfluorodioxole.

20 microliters of 6.5% by weight perfluoropropionylperoxide dissolved in $\text{CCl}_2\text{F-CF}_2\text{Cl}$, 2.95 mmoles of perfluorobisvinylloxymethane and 0.6 mmoles of perfluoro-1,3-dioxole are loaded into a 42 ml glass reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C , evacuated, brought to room temperature and cooled again twice; at the end of this operation sequence the reactor is kept at 40°C for 8 hours under magnetic stirring. The reactor is brought to the temperature of the liquid nitrogen and connected to a vacuum system kept at the pressure of 10^{-3} mbar, it is then let to reach room temperature fractionating the vapours by traps cooled at -90°C , -120°C , and -196°C .

The trap at -90°C contains 1.92 mmoles of bisvinylloxymethane and $\text{CFCl}_2\text{-CF}_2\text{Cl}$ used as solvent. The trap at -120°C contains 0.22 mmoles of perfluorodioxole and 0.01 mmoles of perfluorobisvinylloxymethane. The trap at -196°C contains 0.1 mmoles of perfluorodioxole.

After distillation of the solvent and of the unreacted monomers and stripping of the polymer under vacuum at the temperature of 120°C for 2 hours, 280 mg of polymer are isolated.

The weight balance determined by gas chromatography analysis of the content of the traps containing the unreacted monomers allows to calculate the amount of bisvinylloxymethane in the polymer, which results to be 78% by moles.

The polymer T_g determined by DSC is 94.5°C . The DSC graph does not show any melting endotherm, therefore the polymer is amorphous. The TGA shows a weight loss of 10% at 494°C .

EXAMPLE 31 Copolymer of bis(2-fluorovinylloxy)-difluoromethane with vinylidene fluoride.

1 ml of 6.5% by weight perfluoropropionylperoxide dissolved in $\text{CCl}_2\text{F-CF}_2\text{Cl}$, 10 ml of $\text{CCl}_2\text{F-CF}_2\text{Cl}$ and 1.5 g of bis(2-fluorovinylloxy)-difluoromethane (m.w.=172) are loaded into a 40 ml AISI-316 reactor for polymerization, equipped with magnetic stirring and opening for the loading and unloading of the reactants.

The so loaded reactor is cooled to -196°C , evacuated, brought to room temperature and cooled again twice; at the end of this operation sequence the reactor is kept at 30°C and pressurized with vinylidene fluoride up to 20 atm. When the total pressure drops to 18 atm the reaction is blocked and gaseous reagents are removed at atmospheric pressure.

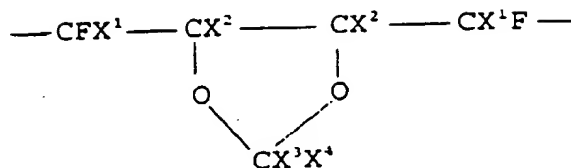
After distillation of the solvent and of the unreacted monomers and stripping of the polymer under vacuum at the temperature of 80°C for 2 hours, 310 mg of polymer are isolated.

The polymer T_g determined by DSC is 39.9°C . The DSC graph does not show any melting endotherm, therefore the polymer is amorphous. The TGA shows a weight loss of 6% at 300°C .

Claims

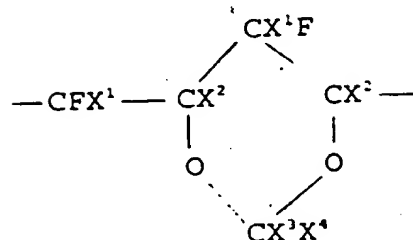
1. Polymers and copolymers of one or more fluorinated dienes of structure $\text{CFX}^1=\text{CX}^2\text{-O-CX}^3\text{X}^4\text{-O-CX}^2=\text{CX}^1\text{F}$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H, and X^3 and X^4 , equal to or different from each other, are F or CF_3 , said dienes essentially forming cyclic repetitive units having the structures:

a)

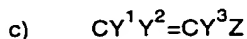


and

b)

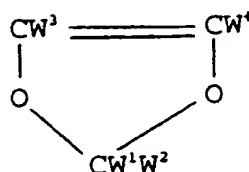


wherein X^1 , X^2 , X^3 and X^4 have the meaning indicated above, at least one of the comonomers used in preparing said copolymers being a compound with an ethylene unsaturation having the structure:



wherein Y^1 , Y^2 , Y^3 , equal to or different from each other, are F, Cl or H and Z is F, Cl, H, R^f or $O-R^f$ with R^f perfluoroalkyl radical having from one to five carbon atoms, the compound containing at most 2 chlorine atoms; or alkylvinylethers of said structure c) wherein Z is $O-R^f$ with R^f equal to a perfluoroalkyl radical having from 1-5 carbon atoms, with the same limitations relating to the content of chlorine atoms; or

d)



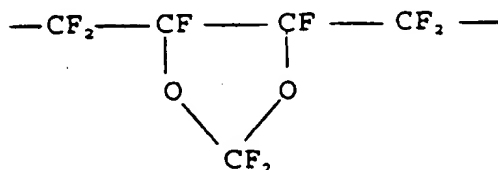
wherein W^1 and W^2 , equal to or different from each other, represent F or CF_3 , W^3 represents F or H, and W^4 represents F, H, R^f or $O-R^f$ with R^f perfluoroalkyl radical containing from 1 to 5 carbon atoms.

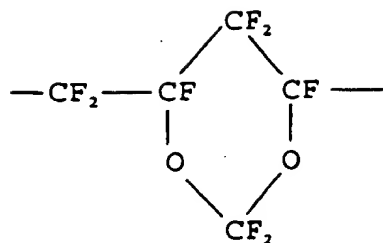
2. Polymers and copolymers according to claim 1 wherein $X^2=X^1$.

3. Polymers and copolymers according to claim 2 wherein $X^2=X^1=F$.

4. Polymers and copolymers according to claim 3, wherein the fluorinated diene is $CF_2=CF-O-CF_2-O-CF=CF_2$ and the cyclic repetitive units derived from it have the following structures:

a)





b)

5. Copolymers according to anyone of the claims from 1 to 4 comprising a comonomer selected from: tetrafluoroethylene, vinylidene fluoride, ethylene, chlorotrifluoroethylene, trifluoroethylene, perfluoromethylvinylether, perfluoropropylvinylether, perfluoropropene, perfluoro 2,2-dimethyl-1,3-dioxole, 1,3-dioxole, perfluoro 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole.

6. Copolymers according to claim 5 wherein the comonomer is tetrafluoroethylene.

7. Copolymers according to claim 5 wherein the comonomer is vinylidene fluoride.

8. Copolymers according to claim 5 wherein the comonomer is selected from perfluoro 1,3-dioxole, perfluoro 2,2-dimethyl-1,3-dioxole and 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole.

9. Fluorinated derivatives of bisvinylloxymethane of structure $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2'=\text{CX}^1'\text{F}$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 .

10. Fluorinated derivatives of bisvinylloxymethane of structure $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2'=\text{CX}^1'\text{F}$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 ; X^1' and X^2' , equal to or different from each other, are X^1 or X^2 , with the proviso that when X^1 is different from X^2 also X^1' is different from X^2' .

11. Fluorinated derivatives of bisvinylloxymethane of structure $\text{CFX}^1=\text{CX}^1'-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^1'=\text{CX}^1'\text{F}$ wherein X^1 is F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 .

12. Fluorinated derivatives of bisvinylloxymethane of structure $\text{CF}_2=\text{CF}-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CF}=\text{CF}_2$, wherein X^3 and X^4 , equal to or different from each other, are F or CF_3 .

13. Perfluorobisvinylloxymethane $\text{CF}_2=\text{CF}-\text{O}-\text{CF}_2-\text{O}-\text{CF}=\text{CF}_2$.

14. Process for the preparation of bisvinylloxymethane derivatives having the structure $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2'=\text{CX}^1'\text{F}$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 ; X^1' and X^2' , equal to or different from each other, are X^1 or X^2 , with the proviso that if X^1 is different from X^2 also X^1' is to be different from X^2' , comprising:

- i) addition of an olefin of formula $\text{CX}^1\text{X}^5=\text{CX}^2\text{X}^6$, wherein X^1 , X^2 , X^5 and X^6 equal to or different from each other are F, Cl, H or Br, the Br atoms being 2 at most and in such a case being bound to different carbon atoms, X^1 and X^5 not being both F, X^2 and X^6 not being both F, X^1 and X^2 being F only if X^5 and X^6 are different from F and not being both H, in a reactor where an hypofluorite of general formula $\text{CX}^3\text{X}^4(\text{OF})_2$ is essentially always present, where X^3 and X^4 equal to or different from each other, are F or CF_3 , dissolved in an inert solvent and having a concentration between 0.001 M and 10 M, at a temperature from -140°C to $+60^\circ\text{C}$;
- ii) separation of the reaction product of two olefin molecules and one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.
- iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

15. Process for preparing bisvinylloxymethane derivatives having the structure $\text{CFX}^1=\text{CX}^1'-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^1'=\text{CX}^1'\text{F}$, wherein X^1 is F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 , which comprises:

i) addition of an olefin of formula $CX^1X^5=CX^1X^6$, wherein X^1 , X^5 and X^6 equal to or different from each other, are F, Cl, Br or H, X^1 and X^5 not being both F, X^1 and X^6 not being both F, X^1 being Br only if X^5 and X^6 equal to each other are different from Br, X^1 being F only if X^5 and X^6 are different from F and not both H, to an hypofluorite of general formula $CX^3X^4(OF)_2$ wherein X^3 and X^4 equal to or different from each other, are F or CF_3 , dissolved in an inert solvent having a concentration between 0.001 M and 10 M, at a temperature from $-140^\circ C$ to $+60^\circ C$;

ii) separation of the reaction product of two olefin molecules and one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

16. Process for preparing bisvinylloxymethane derivatives having the structure $CF_2=CF-O-CX^3X^4-O-CF=CF_2$, wherein X^3 and X^4 , equal to or different from each other, are F or CF_3 , which comprises:

i) addition of an olefin of formula $CFX^5=CFX^6$, wherein X^5 is Cl, Br or H, and X^6 is Cl or Br, to an hypofluorite of general formula $CX^3X^4(OF)_2$ wherein X^3 and X^4 , equal to or different from each other, are F or CF_3 , dissolved in an inert solvent having a concentration between 0.001 M and 10 M, at a temperature from $-140^\circ C$ to $+60^\circ C$;

ii) separation of the reaction product of two olefin molecules and one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

17. Process according to anyone of the claims from 14 to 16 wherein the solution of the hypofluorite has a concentration from 0.1 M to 4 M.

18. Process according to anyone of the claims from 14 to 16 wherein the reaction temperature is from $-120^\circ C$ to $0^\circ C$.

19. Process according to anyone of the claims from 14 to 18, characterized in that the solution of hypofluorite is prepared using an inert solvent of polar type.

20. Process according to claim 19 characterized in that the solvent is selected from hydrogenfluorocarbons, hydrogenchlorocarbons, fluorochlorocarbons, hydrogenfluorocarbons trifluoroacetic acid, trifluoroacetic anhydride, acetic nitrile, hydrofluoric acid, sulphur dioxide, trifluoromethanesulphonic acid, $CF_2Cl-CFCl-SO_2F$, mixtures thereof and mixtures of one or more of the same with a solvent having low polarity.

21. Process according to claim 14 wherein the olefin is selected from: $CFCI=CFCl$, $CHCl=CHCl$, $CHCl=CCl_2$, $CCl_2=CCl_2$, $CH_2=CF_2$, $CF_2=CF_2$, $CFH=CFCl$, $CFCI=CHCl$, $CH_2=CCl_2$, $CH_2=CFCI$.

22. Process for the preparation of bisvinylloxymethane derivatives having the structure $CFX^1=CX^2-O-CX^3X^4-O-CX^2=CX^1F$, wherein X^1 and X^2 , equal to or different from each other, are F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 ; X^1 and X^2 , equal to or different from each other, are X^1 or X^2 , with the proviso that if X^1 is different from X^2 also X^1 is to be different from X^2 , which comprises:

i) addition of hypofluorite of general formula $CX^3X^4(OF)_2$ wherein X^3 and X^4 equal to or different from each other, are F or CF_3 to an olefin of formula

$CX^1X^5=CX^2X^6$, wherein X^1 , X^2 , X^5 and X^6 , equal to or different from each other, are F, Cl, H or Br; the Br atoms being 2 at most and in such a case bound to different carbon atoms; X^1 and X^5 not being both F; X^2 and X^6 not being both F; X^1 and X^2 being F only if X^5 and X^6 are different from F and not being both H, said olefin being dissolved in an inert solvent of polar type, at a temperature from $-140^\circ C$ to $+60^\circ C$;

ii) separation of the reaction product of two olefin molecules with one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

23. Process for the preparation of bisvinylloxymethane derivatives having the structure $CFX^1=CX^1-O-CX^3X^4-O-CX^1=CX^1F$, wherein X^1 is F, Cl or H; X^3 and X^4 , equal to or different from each other, are F or CF_3 , comprising:

i) addition of hypofluorite of general formula $CX^3X^4(OF)_2$ wherein X^3 and X^4 , equal to or different from each

other, are F or CF₃ to an olefin of formula CX¹X⁵=CX¹X⁶, wherein X¹, X⁵ and X⁶, equal to or different from each other, are F, Cl, Br or H; X¹ and X⁵ not being both F; X¹ and X⁶ not being both F; X¹ being Br only if X⁵ and X⁶ equal to each other are different from Br, X¹ being F only if X⁵ and X⁶ are different from F and not being both H, said olefin being dissolved in an inert solvent of polar type, at a temperature from -140°C to +60°C;

ii) separation of the reaction product of two olefin molecules with one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

24. Process for the preparation of bisvinylloxymethane derivatives having the structure CF₂=CF-O-CX³X⁴-O-CF=CF₂, wherein X³ and X⁴, equal to or different from each other, are F or CF₃, comprising:

i) addition of hypofluorite of general formula CX³X⁴(OF)₂ wherein X³ and X⁴ equal to or different from each other, are F or CF₃ to an olefin of formula CFX⁵=CFX⁶, wherein X⁵ is Cl, Br or H, and X⁶ is Cl or Br, dissolved in an inert solvent of polar type, at a temperature from -140°C to +60°C;

ii) separation of the reaction product of two olefin molecules with one hypofluorite molecule from the reaction mixture obtained in i) by fractional distillation.

iii) dehalogenation or dehydrohalogenation of the product obtained in ii), wherein the eliminated halogen atoms are Cl or Br.

25. Process according to anyone of the claims from 22 to 24 wherein the reaction temperature is from -120°C to 0°C.

26. Process according to anyone of the claims from 22 to 25 characterized in that the solvent is selected from hydrogenfluorocarbons, hydrogenchlorocarbons, fluorochlorocarbons, hydrogenchlorofluorocarbons, trifluoroacetic acid, trifluoroacetic anhydride, acetic nitrile, hydrofluoric acid, sulphur dioxide, trifluoromethaneacid, CF₂Cl-CFCl-SO₂F, mixtures thereof and mixtures of one or more of the same with a solvent having low polarity.

27. Process according to claim 22 wherein the olefin is selected from: CFCl=CFCl, CHCl=CHCl, CHCl=CCl₂, CCl₂=CCl₂, CH₂=CF₂, CF₂=CF₂, CFH=CFCl, CFCl=CHCl, CH₂=CCl₂, CH₂=CFCl.

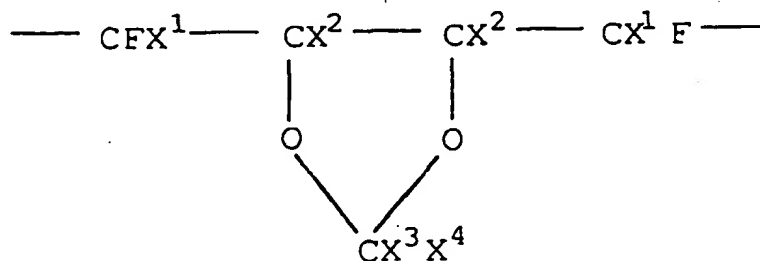
28. Fluorinated derivatives of bisethoxymethane obtainable as intermediates after the steps i) and ii) of the processes of claims 14 and 22.

29. Use of polymers and copolymers according to anyone of the claims from 1 to 8 in the preparation of coating compositions.

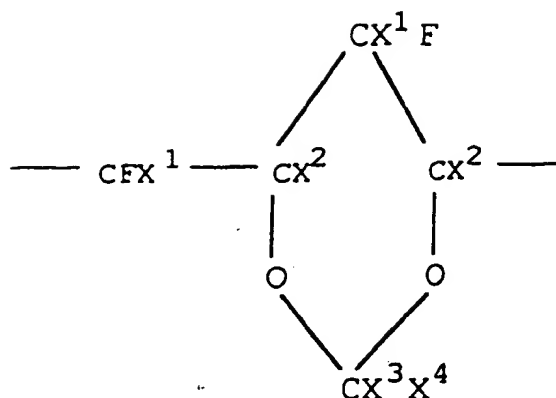
30. Use of fluorinated compounds according to claim 28 as solvents with low environmental impact.

Patentansprüche

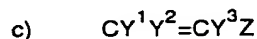
1. Polymere und Copolymere von einem oder mehreren fluorierten Dienen der Struktur CFX¹=CX²-O-CX³X⁴-O-CX²=CX¹F, wobei X¹ und X² gleich oder voneinander verschieden F, Cl oder H sind, und X³ und X⁴ gleich oder voneinander verschieden F oder CF₃ sind, wobei diese Diene im wesentlichen cyclische sich wiederholende Einheiten mit den Strukturen:



und

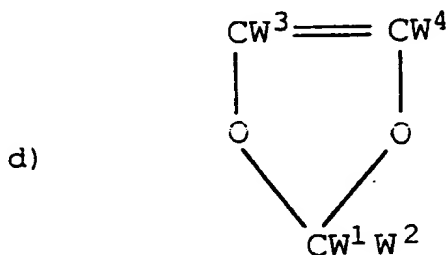


20 bilden, wobei X¹, X², X³ und X⁴ die oben aufgeführte Bedeutung haben, wobei mindestens eines der zur Herstellung der Copolymeren verwendeten Comonomeren eine Verbindung mit einer Ethylen-Ungesättigkeit mit der folgenden Struktur:



ist, wobei Y¹, Y², Y³ gleich oder voneinander verschieden F, Cl oder H sind, und Z gleich F, Cl, H, R¹ oder O-R¹ ist, wobei R¹ ein Perfluoralkylrest mit 1 bis 5 Kohlenstoffatomen ist, wobei die Verbindung höchstens 2 Chlortome enthält; oder

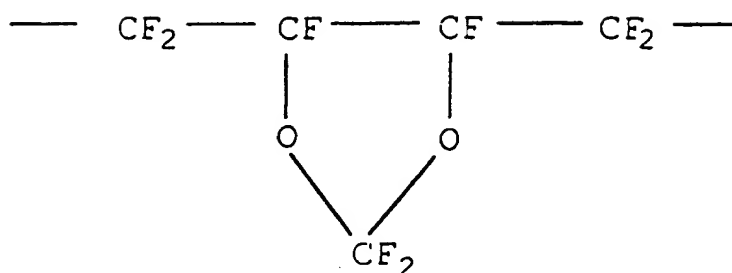
30 ein Alkylvinylether der Struktur c) ist, wobei Z gleich O-R¹ ist, mit R¹ gleich einem Perfluoralkylrest mit 1 - 5 Kohlenstoffatomen mit denselben Einschränkungen hinsichtlich des Gehalts an Chlortomen; oder



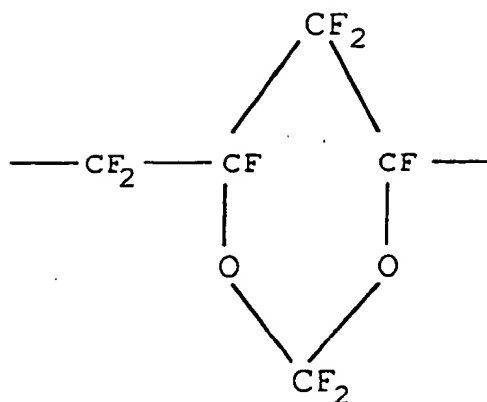
40 wobei W¹ und W² gleich oder voneinander verschieden F oder CF₃ darstellen, W³ für F oder H steht, und W⁴ für F, H, R¹ oder O-R¹ steht, wobei R¹ ein Perfluoralkylrest mit 1 bis 5 Kohlenstoffatomen ist.

2. Polymere und Copolymere gemäß Anspruch 1, wobei X²=X¹.
3. Polymere und Copolymere gemäß Anspruch 2, wobei X²=X¹=F.
4. Polymere und Copolymere gemäß Anspruch 3, wobei das fluorierte Dien CF₂=CF-O-CF₂-O-CF=CF₂ ist, und die davon abgeleiteten cyclischen sich wiederholenden Einheiten die folgenden Strukturen besitzen:

a)



b)



5. Copolymere gemäß einem der vorhergehenden Ansprüche 1 bis 4, welche ein Comonomer umfassen, ausgewählt aus: Tetrafluorethylen, Vinylidenfluorid, Ethylen, Chlortrifluorethylen, Trifluorethylen, Perfluormethylvinylether, Perfluorpropylvinylether, Perfluorpropen, Perfluor-2,2-dimethyl-1,3-dioxol, Perfluor-1,3-dioxol, 2,2,4-Trifluor-5-trifluormethoxy-1,3-dioxol.
6. Copolymere gemäß Anspruch 5, wobei das Comonomer Tetrafluorethylen ist.
7. Copolymere gemäß Anspruch 5, wobei das Comonomer Vinylidenfluorid ist.
8. Copolymere gemäß Anspruch 5, wobei das Comonomer ausgewählt ist aus Perfluor-1,3-dioxol, Perfluor-2,2-dimethyl-1,3-dioxol und 2,2,4-Trifluor-5-trifluormethoxy-1,3-dioxol.
9. Fluorierte Derivate eines Bisvinylloxymethans der Struktur $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2=\text{CX}^1\text{F}$, wobei X^1 und X^2 gleich oder voneinander verschieden F, Cl oder H sind; X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind.
10. Fluorierte Derivate eines Bisvinylloxymethans der Struktur $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2=\text{CX}^1\text{F}$, wobei X^1 und X^2 gleich oder voneinander verschieden F, Cl oder H sind; X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind; X^1 und X^2 gleich oder voneinander verschieden X^1 oder X^2 sind, unter der Voraussetzung, daß wenn X^1 verschieden von X^2 ist, auch X^1 verschieden von X^2 ist.
11. Fluorierte Derivate eines Bisvinylloxymethans der Struktur $\text{CFX}^1=\text{CX}^1-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^1=\text{CX}^1\text{F}$, wobei X^1 gleich F, Cl oder H ist; X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind.
12. Fluorierte Derivate eines Bisvinylloxymethans der Struktur $\text{CF}_2=\text{CF}-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CF}=\text{CF}_2$, wobei X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind.
13. Perfluorbisvinylloxymethan $\text{CF}_2=\text{CF}-\text{O}-\text{CF}_2-\text{O}-\text{CF}=\text{CF}_2$.
14. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten der Struktur $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2=\text{CX}^1\text{F}$, wo-

bei X^1 und X^2 gleich oder voneinander verschieden F, Cl oder H sind, und X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind; X^1 und X^2 gleich oder voneinander verschieden X^1 oder X^2 sind, unter der Voraussetzung, daß wenn X^1 verschieden von X^2 ist, auch X^1 verschieden von X^2 ist, welches Verfahren folgende Schritte umfaßt:

i) Zugabe eines Olefins der Formel $CX^1X^5=CX^2X^6$, wobei X^1 , X^2 , X^5 und X^6 gleich oder voneinander verschieden F, Cl, H oder Br sind, wobei maximal 2 Br-Atome vorhanden sind und in einem solchen Fall an verschiedene Kohlenstoffatome gebunden sind, X^1 und X^5 nicht beide F sind, X^2 und X^6 nicht beide F sind, X^1 und X^2 lediglich F sind, wenn X^5 und X^6 verschieden von F und nicht beide H sind, in einen Reaktor, in welchem im wesentlichen immer ein Hypofluorit der allgemeinen Formel $CX^3X^4(OF)_2$ vorhanden ist, wobei X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind, gelöst in einem inerten Lösungsmittel und mit einer Konzentration von zwischen 0,001 M und 10 M, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen und einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung der in ii) erhaltenen Produkte, wobei die eliminierten Halogenatome Cl oder Br sind.

15. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten mit der Struktur $CFX^1=CX^1-O-CX^3X^4-O-CX^1=CX^1F$, wobei X^1 gleich F, Cl oder H ist; X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind, welches folgendes umfaßt:

i) Zugabe eines Olefins der Formel $CX^1X^5=CX^1X^6$, wobei X^1 , X^5 und X^6 gleich oder voneinander verschieden F, Cl, Br oder H sind, X^1 und X^5 nicht beide F sind, X^1 und X^6 nicht beide F sind, X^1 lediglich Br ist, wenn X^5 und X^6 miteinander gleich und von Br verschieden sind, X^1 lediglich F ist, wenn X^5 und X^6 verschieden von F und nicht beide H sind, zu einem Hypofluorit der allgemeinen Formel $CX^3X^4(OF)_2$, wobei X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind, gelöst in einem inerten Lösungsmittel mit einer Konzentration zwischen 0,001 M und 10 M, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen und einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung des in ii) erhaltenen Produkts, wobei die eliminierten Halogenatome Cl oder Br sind.

16. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten mit der Struktur $CF_2=CF-O-CX^3X^4-O-CF=CF_2$, wobei X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind, welches folgendes umfaßt:

i) Zugabe eines Olefins der Formel $CFX^5=CFX^6$, wobei X^5 gleich Cl, Br oder H ist, und X^6 gleich Cl oder Br ist, zu einem Hypofluorit der allgemeinen Formel $CX^3X^4(OF)_2$, wobei X^3 und X^4 gleich oder voneinander verschieden F oder CF_3 sind, gelöst in einem inerten Lösungsmittel mit einer Konzentration zwischen 0,001 M und 10 M, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen und einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung des in ii) erhaltenen Produkts, wobei die eliminierten Halogenatome Cl oder Br sind.

17. Verfahren gemäß einem der Ansprüche 14 bis 16, wobei die Lösung des Hypofluorits eine Konzentration von 0,1 M bis 4 M besitzt.

18. Verfahren gemäß einem der Ansprüche 14 bis 16, wobei die Reaktionstemperatur von -120°C bis 0°C reicht...

19. Verfahren gemäß einem der Ansprüche 14 bis 18, dadurch gekennzeichnet, daß die Lösung des Hypofluorits unter Verwendung eines inerten Lösungsmittels vom polaren Typ hergestellt wird.

20. Verfahren gemäß Anspruch 19, dadurch gekennzeichnet, daß das Lösungsmittel ausgewählt ist aus Fluorkohlenwasserstoffen, Chlorkohlenwasserstoffen, Fluorchlorkohlenstoffen, Fluorkohlenwasserstoffen, Trifluoressigsäure, Trifluoressigsäureanhydrid, Acetonitril, Fluorwasserstoffsäure, Schwefeldioxid, Trifluormethansulfonsäure, $\text{CF}_2\text{Cl-CFCl-SO}_2\text{F}$, deren Mischungen und Mischungen von einem oder mehreren derartigen mit einem Lösungsmittel mit geringer Polarität.

21. Verfahren gemäß Anspruch 14, wobei das Olefin ausgewählt ist aus: CFCl=CFCl , CHCl=CHCl , CHCl=CCl_2 , $\text{CCl}_2=\text{CCl}_2$, $\text{CH}_2=\text{CF}_2$, $\text{CF}_2=\text{CF}_2$, CFH=CFCl , CFCl=CHCl , $\text{CH}_2=\text{CCl}_2$, $\text{CH}_2=\text{CFCl}$.

22. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten mit der Struktur $\text{CFX}^1=\text{CX}^2-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^2=\text{CX}^1\text{F}$, wobei X^1 und X^2 gleich oder voneinander verschiedenen F, Cl oder H sind, X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind; X^1 und X^2 gleich oder voneinander verschiedenen X^1 oder X^2 sind, unter der Voraussetzung, daß wenn X^1 verschieden von X^2 ist, auch X^1 verschieden von X^2 ist, welches Verfahren folgende Schritte umfaßt:

i) Zugeben eines Hypofluorits der allgemeinen Formel $\text{CX}^3\text{X}^4(\text{OF})_2$, wobei X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind, zu einem Olefin der Formel $\text{CX}^1\text{X}^5=\text{CX}^2\text{X}^6$, wobei X^1 , X^2 , X^5 und X^6 gleich oder voneinander verschiedenen F, Cl, H oder Br sind; wobei maximal 2 Br-Atome vorhanden sind und in einem solchen Fall an verschiedene Kohlenstoffatome gebunden sind, X^1 und X^5 nicht beide F sind, X^2 und X^6 nicht beide F sind, X^1 und X^2 lediglich F sind, wenn X^5 und X^6 verschieden von F und nicht beide H sind, wobei das Olefin in einem inerten Lösungsmittel vom polaren Typ gelöst ist, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen mit einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung des in ii) erhaltenen Produkts, wobei die eliminierten Halogenatome Cl oder Br sind.

23. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten mit der Struktur $\text{CFX}^1=\text{CX}^1-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CX}^1=\text{CX}^1\text{F}$, wobei X^1 gleich F, Cl oder H ist; X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind, welches folgendes umfaßt:

i) Zugeben eines Hypofluorits der allgemeinen Formel $\text{CX}^3\text{X}^4(\text{OF})_2$, wobei X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind, zu einem Olefin der Formel $\text{CX}^1\text{X}^5=\text{CX}^1\text{X}^6$, wobei X^1 , X^5 und X^6 gleich oder voneinander verschiedenen F, Cl, Br oder H sind, X^1 und X^5 nicht beide F sind, X^1 und X^6 nicht beide F sind, X^1 lediglich Br ist, wenn X^5 und X^6 miteinander gleich und von Br verschieden sind, X^1 lediglich F ist, wenn X^5 und X^6 verschieden von F und nicht beide H sind, wobei das Olefin in einem inerten Lösungsmittel vom polaren Typ gelöst ist, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen mit einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung des in ii) erhaltenen Produkts, wobei die eliminierten Halogenatome Cl oder Br sind.

24. Verfahren zur Herstellung von Bisvinylloxymethan-Derivaten mit der Struktur $\text{CF}_2=\text{CF}-\text{O}-\text{CX}^3\text{X}^4-\text{O}-\text{CF}=\text{CF}_2$, wobei X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind, welches folgendes umfaßt:

i) Zugeben eines Hypofluorits der allgemeinen Formel $\text{CX}^3\text{X}^4(\text{OF})_2$, wobei X^3 und X^4 gleich oder voneinander verschiedenen F oder CF_3 sind, zu einem Olefin der Formel $\text{CFX}^5=\text{CFX}^6$, wobei X^5 gleich Cl, Br oder H ist, und X^6 gleich Cl oder Br ist, gelöst in einem inerten Lösungsmittel vom polaren Typ, bei einer Temperatur von -140°C bis $+60^\circ\text{C}$;

ii) Trennen des Umsetzungsprodukts von zwei Olefinmolekülen mit einem Hypofluoritmolekül von der in i) erhaltenen Reaktionsmischung durch fraktionierte Destillation;

iii) Dehalogenierung oder Dehydrohalogenierung des in ii) erhaltenen Produkts, wobei die eliminierten Halo-

genatome Cl oder Br sind.

25. Verfahren gemäß einem der Ansprüche 22 bis 24, wobei die Reaktionstemperatur von -120°C bis 0°C reicht.

26. Verfahren gemäß einem der Ansprüche 22 bis 25, dadurch gekennzeichnet, daß das Lösungsmittel ausgewählt ist aus Fluorkohlenwasserstoffen, Chlorkohlenwasserstoffen, Fluorchlorkohlenstoffen, Chlorfluorkohlenwasserstoffen, Trifluoressigsäure, Trifluoressigsäureanhydrid, Acetonitril, Fluorwasserstoffsäure, Schwefeldioxid, Trifluormethansulfonsäure, $\text{CF}_2\text{Cl-CFCl-SO}_2\text{F}$, deren Mischungen und Mischungen von einem oder mehreren derartigen mit einem Lösungsmittel mit geringer Polarität.

27. Verfahren gemäß Anspruch 22, wobei das Olefin ausgewählt ist aus: CFCl=CFCl , CHCl=CHCl , CHCl=CCl_2 , $\text{CCl}_2=\text{CCl}_2$, $\text{CH}_2=\text{CF}_2$, $\text{CF}_2=\text{CF}_2$, CFH=CFCl , CFCl=CHCl , $\text{CH}_2=\text{CCl}_2$, $\text{CH}_2=\text{CFCl}$.

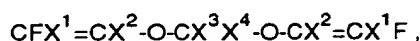
28. Fluorierte Derivate von Bisethoxymethan, welche als Zwischenprodukte nach den Schritten i) und ii) der Verfahren der Ansprüche 14 und 22 erhältlich sind.

29. Verwendung von Polymeren und Copolymeren gemäß einem der Ansprüche 1 bis 8 bei der Herstellung von Beschichtungszusammensetzungen.

30. Verwendung von fluorierten Verbindungen gemäß Anspruch 28 als Lösungsmittel mit geringen Auswirkungen auf die Umwelt.

Revendications

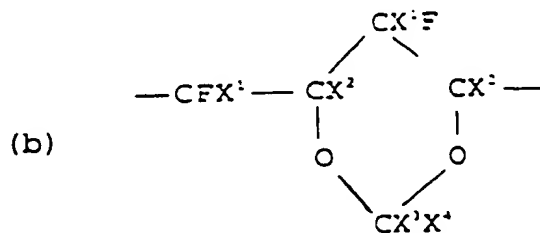
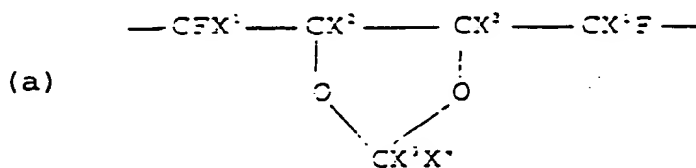
1. Polymères et copolymères d'un ou plusieurs diènes fluorés de structure :



dans laquelle :

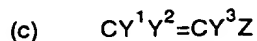
- X^1 et X^2 , identiques entre eux ou différents l'un de l'autre, représentent F, Cl ou H ; et
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ,

lesdits diènes formant essentiellement des unités répétitives cycliques ayant les structures :



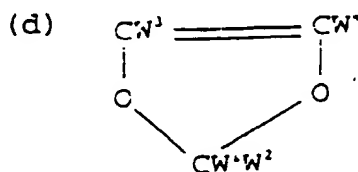
dans lesquelles X^1 , X^2 , X^3 et X^4 ont la signification indiquée ci-dessus,

au moins l'un des comonomères utilisés dans la préparation desdits copolymères étant un composé avec une insaturation éthylénique, ayant la structure :



dans laquelle :

- Y^1, Y^2, Y^3 , identiques entre eux ou différents les uns des autres, représentent F, Cl ou H ; et
- Z représente F, Cl, H, R^f ou $O-R^f$ avec R^f radical perfluoroalkyle ayant de un à cinq atomes de carbone, le composé contenant au plus 2 atomes de chlore ; ou des alkylvinyléthers de ladite structure (c) dans laquelle Z représente $O-R^f$ avec R^f égal à un radical perfluoroalkyle ayant de 1 à 5 atomes de carbone, avec les mêmes limitations concernant la teneur en atomes de chlore ; ou



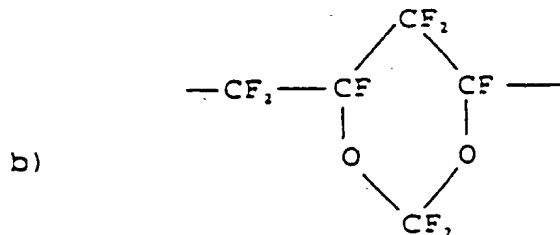
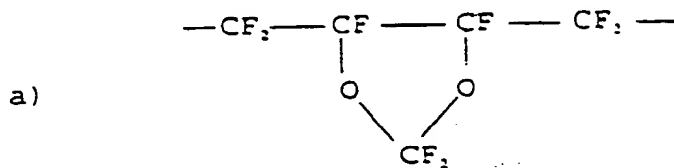
dans laquelle :

- W^1 et W^2 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ;
- W^3 représente F ou H ; et
- W^4 représente F, H, R^f ou $O-R^f$ avec R^f radical perfluoroalkyle contenant de 1 à 5 atomes de carbone.

2. Polymères et copolymères selon la revendication 1, dans lesquels $X^2 = X^1$.

3. Polymères et copolymères selon la revendication 2, dans lesquels $X^2 = X^1 = F$.

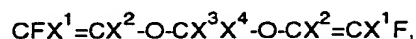
4. Polymères et copolymères selon la revendication 3, dans lesquels le diène fluoré représente $CF_2=CF-O-CF_2-O-CF=CF_2$ et les unités répétitives cycliques provenant de celui-ci ont les structures suivantes :



5. Copolymères selon l'une quelconque des revendications 1 à 4, comprenant un comonomère choisi parmi : le

tétrafluoroéthylène, le fluorure de vinylidène, l'éthylène, le chlorotrifluoroéthylène, le trifluoroéthylène, le perfluorométhylvinyléther, le perfluoropropylvinyléther, le perfluoropropène, le perfluoro 2,2-diméthyl-1,3-dioxole, le perfluoro 1,3-dioxole, le 2,2,4-trifluoro-5-trifluorométhoxy-1,3-dioxole,

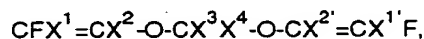
- 5 6. Copolymères selon la revendication 5, dans lesquels le comonomère est le tétrafluoroéthylène.
7. Copolymères selon la revendication 5, dans lesquels le comonomère est le fluorure de vinylidène.
8. Copolymères selon la revendication 5, dans lesquels le comonomère est choisi parmi le perfluoro 1,3-dioxole, le perfluoro 2,2-diméthyl-1,3-dioxole et le 2,2,4-trifluoro-5-trifluorométhoxy-1,3-dioxole.
9. Dérivés fluorés de bisvinyloxyméthane de structure :



dans laquelle :

- X^1 et X^2 , identiques entre eux ou différents l'un de l'autre, représentent F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 .

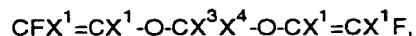
10. Dérivés fluorés de bisvinyloxyméthane de structure :



dans laquelle :

- X^1 et X^2 identiques entre eux ou différents l'un de l'autre, représentent F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ;
- $\text{X}^{1'}$ et $\text{X}^{2'}$, identiques entre eux ou différents l'un de l'autre, représentent X^1 ou X^2 , à la condition que, lorsque X^1 est différent de X^2 , également $\text{X}^{1'}$ est différent de $\text{X}^{2'}$.

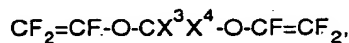
11. Dérivés fluorés de bisvinyloxyméthane de structure :



dans laquelle :

- X^1 représente F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 .

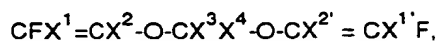
12. Dérivés fluorés de bisvinyloxyméthane de structure :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 .

13. Perfluorobisvinyloxyméthane $\text{CF}_2=\text{CF}-\text{O}-\text{CF}_2-\text{O}-\text{CF}=\text{CF}_2$.

14. Procédé de préparation de dérivés de bisvinyloxyméthane ayant la structure :

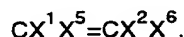


dans laquelle :

- X^1 et X^2 , identiques entre eux ou différents l'un de l'autre, représentent F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ;
- X^1 et X^2 , identiques entre eux ou différents l'un de l'autre, représentent X^1 ou X^2 ,

à la condition que, si X^1 est différent de X^2 , également X^1 doit être différent de X^2 .
comprenant :

(i) l'addition d'une oléfine de formule :

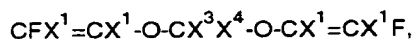


dans laquelle X^1 , X^2 , X^5 et X^6 , identiques entre eux ou différents les uns des autres, représentent F, Cl, H ou Br, les atomes de Br étant de 2 au plus et dans un tel cas étant liés à des atomes de carbone différents, X^1 et X^5 ne représentant pas tous deux F, X^2 et X^6 ne représentant pas tous deux F, X^1 et X^2 représentant F seulement si X^5 et X^6 sont différents de F et ne représentant pas tous deux H, dans un réacteur où un hypofluorite de formule générale $CX^3X^4(OF)_2$ est pratiquement toujours présent, où X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , dissous dans un solvant inerte et ayant une concentration comprise entre 0,001 M et 10 M, à une température de $-140^\circ C$ à $+60^\circ C$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine et d'une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

(iii) la déshalogénation ou la déshydrohalogénation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés sont Cl ou Br.

15. Procédé de préparation de dérivés de bisvinyloxyméthane ayant la structure :

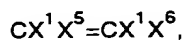


dans laquelle :

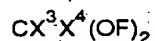
- X^1 représente F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ,

qui comprend :

(i) l'addition d'une oléfine de formule :



dans laquelle X^1 , X^5 et X^6 , identiques entre eux ou différents les uns des autres, représentent F, Cl, Br ou H, X^1 et X^5 ne représentant pas tous deux F, X^1 et X^6 ne représentant pas tous deux F, X^1 représentant Br seulement si X^5 et X^6 identiques entre eux sont différents de Br, X^1 représentant F seulement si X^5 et X^6 sont différents de F et n'étant pas tous deux H, sur un hypofluorite de formule générale :

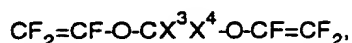


dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , dissous dans un solvant inerte, ayant une concentration comprise entre 0,001 M et 10 M, à une température de $-140^\circ C$ à $+60^\circ C$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine et d'une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

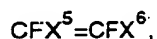
(iii) la déshalogénation ou la déshydrohalogénation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés sont Cl ou Br.

16. Procédé de préparation de dérivés de bisvinyloxyméthane ayant la structure :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , qui comprend :

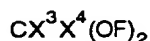
(i) l'addition d'une oléfine de formule :



dans laquelle :

- X^5 représente Cl, Br ou H ; et
- X^6 représente Cl ou Br,

sur un hypofluorite de formule générale :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , dissous dans un solvant inerte, ayant une concentration comprise entre 0,001 M et 10 M, à une température de -140°C à $+60^\circ\text{C}$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine et d'une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

(iii) la déshalogénéation ou la déshydrohalogénéation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés sont Cl ou Br.

17. Procédé selon l'une quelconque des revendications 14 à 16, dans lequel la solution de l'hypofluorite a une concentration de 0,1 M à 4 M.

18. Procédé selon l'une quelconque des revendications 14 à 16, dans lequel la température de réaction est de -120°C à 0°C .

19. Procédé selon l'une quelconque des revendications 14 à 18, caractérisé par le fait que la solution d'hypofluorite est préparée à l'aide d'un solvant inerte de type polaire.

20. Procédé selon la revendication 19, caractérisé par le fait que le solvant est choisi parmi les hydrogénofluorocarbures, les hydrogénochlorocarbures, les fluorochlorocarbures, les hydrogénofluorocarbures, l'acide trifluoroacétique, l'anhydride trifluoroacétique, le nitrile acétique, l'acide fluorhydrique, le dioxyde de soufre, l'acide trifluorométhane sulfonique, $\text{CF}_2\text{Cl}-\text{CFCl}-\text{SO}_2\text{F}$, des mélanges de ceux-ci et des mélanges d'un ou plusieurs de ceux-ci avec un solvant ayant une faible polarité.

21. Procédé selon la revendication 14, dans lequel l'oléfine est choisie parmi : $\text{CFCl}=\text{CFCl}$, $\text{CHCl}=\text{CHCl}$, $\text{CHCl}=\text{CCl}_2$, $\text{CCl}_2=\text{CCl}_2$, $\text{CH}_2=\text{CF}_2$, $\text{CF}_2=\text{CF}_2$, $\text{CFH}=\text{CFCl}$, $\text{CFCl}=\text{CHCl}$, $\text{CH}_2=\text{CCl}_2$, $\text{CH}_2=\text{CFCl}$.

22. Procédé de préparation de dérivés de bisvinyloxyméthane ayant la structure :

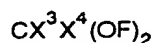


dans laquelle :

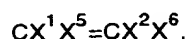
- X^1 et X^2 , égaux entre eux ou différents l'un de l'autre, représentent F, Cl ou H ;
- X^3 et X^4 , égaux entre eux ou différents l'un de l'autre, représentent F ou CF_3 ;
- X^1 et X^2 , identiques entre eux ou différents l'un de l'autre, représentent X^1 ou X^2 ,

avec la condition que, si X^1 est différent de X^2 , également X^1 doit être différent de X^2 , qui comprend :

(i) l'addition d'hypofluorite de formule générale :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , sur une oléfine de formule :

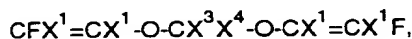


dans laquelle X^1 , X^2 , X^5 et X^6 , identiques entre eux ou différents les uns des autres, représentent F, Cl, H ou Br ; les atomes de Br étant de 2 au plus et dans un tel cas étant liés à des atomes de carbone différents ; X^1 et X^5 ne représentant pas tous deux F ; X^2 et X^6 ne représentant pas tous deux F ; X^1 et X^2 représentant F seulement si X^5 et X^6 sont différents de F et ne représentant pas tous deux H, ladite oléfine étant dissoute dans un solvant inerte de type polaire, à une température de $-140^\circ C$ à $+60^\circ C$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine avec une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

(iii) la déshalogénéation et la déshydrohalogénéation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés sont Cl ou Br.

23. Procédé de préparation de dérivés de bisvinyloxyéthane ayant la structure :

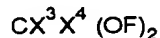


dans laquelle :

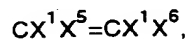
- X^1 représente F, Cl ou H ;
- X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 ,

comprenant :

(i) l'addition d'hypofluorite de formule générale :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 sur une oléfine de formule

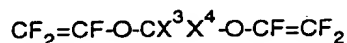


dans laquelle X^1 , X^5 et X^6 , identiques entre eux ou différents les uns des autres, représentent F, Cl, Br ou H ; X^1 et X^5 ne représentant pas tous deux F ; X^1 et X^6 ne représentant pas tous deux F ; X^1 représentant Br seulement si X^5 et X^6 identiques entre eux sont différents de Br, X^1 représentant F seulement si X^5 et X^6 sont différents de F et n'étant pas tous deux H, ladite oléfine étant dissoute dans un solvant inerte de type polaire, à une température de $-140^\circ C$ à $+60^\circ C$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine avec une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

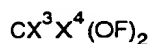
(iii) la déshalogénéation ou la déshydrohalogénéation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés représentent Cl ou Br.

24. Procédé de préparation de dérivés de bisvinyloxyméthane ayant la structure :

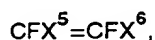


dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , comprenant :

(i) l'addition d'hypofluorite de formule générale :



dans laquelle X^3 et X^4 , identiques entre eux ou différents l'un de l'autre, représentent F ou CF_3 , sur une oléfine de formule :



dans laquelle :

- X^5 représente Cl, Br ou H ; et
- X^6 représente Cl ou Br,

dissous dans un solvant inerte de type polaire, à une température de -140°C à $+60^\circ\text{C}$;

(ii) la séparation du produit de réaction de deux molécules d'oléfine avec une molécule d'hypofluorite à partir du mélange réactionnel obtenu en (i) par distillation fractionnée ;

(iii) la déshalogénéation ou la déshydrohalogénéation du produit obtenu en (ii), dans laquelle les atomes d'halogène éliminés sont Cl ou Br.

25. Procédé selon l'une quelconque des revendications 22 à 24, dans lequel la température de réaction est de -120°C à 0°C .

26. Procédé selon l'une quelconque des revendications 22 à 25, caractérisé par le fait que le solvant est choisi parmi les hydrogénofluorocarbures, les hydrogénochlorocarbures, les fluorochlorocarbures, les hydrogénochlorofluorocarbures, l'acide trifluoroacétique, l'anhydride trifluoroacétique, le nitrile acétique, l'acide fluorhydrique, le dioxyde de soufre, l'acide trifluorométhane sulfonique, $\text{CF}_2\text{Cl}-\text{CFCl}-\text{SO}_2\text{F}$, des mélanges de ceux-ci et des mélanges d'un ou plusieurs de ceux-ci avec un solvant ayant une faible polarité.

27. Procédé selon la revendication 22, dans lequel l'oléfine est choisie parmi : $\text{CFCl}=\text{CFCl}$, $\text{CHCl}=\text{CHCl}$, $\text{CHCl}=\text{CCl}_2$, $\text{CCl}=\text{CCl}_2$, $\text{CH}_2=\text{CF}_2$, $\text{CF}_2=\text{CF}_2$, $\text{CFH}=\text{CFCl}$, $\text{CFCl}=\text{CHCl}$, $\text{CH}_2=\text{CCl}_2$, $\text{CH}_2=\text{CFCl}$.

28. Dérivés fluorés de biséthoxyméthane pouvant être obtenus comme intermédiaires après les étapes (i) et (ii) des procédés des revendications 14 et 22.

29. Utilisation des polymères et copolymères tels que définis à l'une quelconque des revendications 1 à 8, dans la préparation de compositions de revêtement.

30. Utilisation de composés fluorés selon la revendication 28 comme solvants ayant un faible impact sur l'environnement.